Heterologous expression, purification and structural features of native *Dictyostelium discoideum* dye-decolorizing peroxidase bound to a natively incorporated heme

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12   **Keywords: Biocatalysis, Dye-decolorizing peroxidases, Heme incorporation, Lignin degradation,**
13   **Polycyclic dyes, structure enzymology, redox catalysis.**

14   **Abstract**

15   The *Dictyostelium discoideum* dye-decolorizing peroxidase (*Dd*DyP) is a newly discovered
16   peroxidase, which belongs to a unique class of heme peroxidase family that lacks homology to the
17   known members of plant peroxidase superfamily. *Dd*DyP catalyzes the H_2O_2-dependent oxidation of
18   a wide-spectrum of substrates ranging from polycyclic dyes to lignin biomass, holding promise for
19   potential industrial and biotechnological applications. To study the molecular mechanism of *Dd*DyP,
20   highly pure and functional protein with a natively incorporated heme is required, however, obtaining
21   a functional DyP-type peroxidase with a natively bound heme is challenging and often requires addition
22   of expensive biosynthesis precursors. Alternatively, a heme *in vitro* reconstitution approach followed
23   by a chromatographic purification step to remove the excess heme is often used. Here, we show that
24   expressing the *Dd*DyP peroxidase in 2×YT enriched medium at low temperature (20 °C), without
25   adding heme supplement or biosynthetic precursors, allows for a correct native incorporation of heme
26   into the apo-protein, giving rise to a stable protein with a strong Soret peak at 402 nm. Further, we
27   crystallized and determined the native structure of *Dd*DyP at a resolution of 1.95 Å, which verifies the
28   correct heme binding and its geometry. The structural analysis also reveals a binding of two water
29   molecules at the distal site of heme plane bridging the catalytic residues (Arg239 and Asp149) of the
30   GXXDG motif to the heme-Fe(III) via hydrogen bonds. Our results provide new insights into the
31   geometry of native *Dd*DyP active site and its implication on DyP catalysis.

32   **1   Introduction**

33   The *Dictyostelium discoideum* dye-decolorizing peroxidase (*Dd*DyP) is a newly discovered heme
34   peroxidase (Sugano et al., 2007). *Dd*DyP belongs to a new class of DyP-type peroxidases (EC
35   1.11.1.19), which is different from any other known peroxidases (Kim et al., 1999; Sugano et al., 1999;

36 Sugano et al., 2007; Shrestha et al., 2016). This unique peroxidase family has been shown to perform
37 both H_2O_2-dependent oxidation and hydrolytic functions against a wide-spectrum of substrates,
38 ranging from polycyclic dyes, phenolic compounds, sulfides, carotenoids and interestingly lignin
39 biomass, making it a potential candidate for industrial and biotechnological applications including its
40 possible application as bioenergy catalysts as well as biosurfactants in the biodegradation and
41 biotransformation of emerging environmental contaminants (Gan et al., 2022; Rai et al., 2021;
42 Salvachua et al., 2013; Sugano et al., 2021; Xu et al., 2021). This broad substrate specificity is
43 attributed to their unique sequence identity and structural properties (Kim et al., 1999; Yoshida and
44 Sugano, 2023). *Dd*DyP, as other peroxidases, has been found to function in a wide range of pH milieu
45 displaying higher activity at acidic pH with optimal turnover at pH 4.0 and temperatures ranging from
46 20 to 40 °C (Rai et al., 2021; Xu et al., 2021; Colpa et al., 2014). It shows optimal activity at pH 3.0
47 against the known DyP-type peroxidase substrate—the anthraquinone-based dye RB4 (Rai et al.,
48 2021).

49 DyP-type peroxidases share a typical catalytic mechanism with other peroxidases, in which they
50 depend on the H_2O_2 in their oxidative catalytic function as illustrated in scheme 1 (Scocozza et al.,
51 2021; Sugano and Yoshida, 2021). The resting state of the enzyme proceeds into compound I
52 intermediate state upon interaction of the heme-Fe(III) with hydrogen peroxide (H_2O_2), an oxidizing
53 substrate, forming an oxoferryl porphyrin π-cationic radical complex [Fe(IV)=O Por$^\bullet$]$^+$—a
54 porphyrinoid based radical (Colpa et al., 2014; Scocozza et al., 2021). The release of an electron from
55 compound I leads to the formation of compound II [Fe(IV)=O]$^+$ intermediate upon reaction with a
56 reducing substrate giving rise to a radical product, in turn compound II relaxes into the resting state
57 when it reacts with more substrates (Chen et al., 2015; Shrestha et al., 2016; Sugano and Yoshida,
58 2021). The radical product can then be transformed into various sub-products through a non-enzymatic
59 radical coupling. The redox potential of the DyP-type peroxidases, ranging from -50 mV to +250 mV,
60 and that of the substrate determines the feasibility of enzyme catalysis. Generally, a typical DyP-
61 peroxidase catalysis involves several redox couplings, namely Fe^{3+}/Fe^{2+}, compound I/Fe^{3+}, compound
62 II/compound I and compound II/Fe^{3+} (Shrestha et al., 2016). DyP peroxidases also mediate the
63 hydrolysis of substrates such as anthraquinone dyes, implying that the DyP-type peroxidases are
64 bifunctional enzymes (Colpa et al., 2014). However, the exact mechanism for DyP-type peroxidases
65 and how they perform oxidation and hydrolysis for such a wide range of substrates of different
66 chemical properties remain unclear (Xu et al., 2021; Sugano and Yoshida, 2021).

67 Despite the importance of the DyP-type peroxidases as mentioned above, their heterologous expression
68 in *Escherichia coli* (*E. coli*) and other expression systems remain challenging. It hampers, for instance,
69 on the large-scale mechanistic investigation owing to the difficulties associated with the biosynthesis
70 and availability of heme *b*, thereby limiting its native incorporation into the apo-proteins (Fiege et al.,
71 2018; Park and Kim, 2021). It was previously shown that *in vitro* reconstitution is needed for obtaining
72 functional *Dd*DyP with the correct heme stoichiometry (Rai et al., 2021). This is an inherently time-
73 consuming process and may result in excess heme and unspecific binding or altering the protein
74 function, making it limited to robust proteins only (Vogel et al., 1999; Denninger et al., 2000; Lemon
75 et al., 2021). Alternatively, heme and iron supplements or heme biosynthetic precursors such as δ-
76 aminolevulinic acid (δ-ALA) can be used during expression, however this is a highly expensive
77 approach as large amounts of such supplements are needed (Fiege et al., 2018).

78 Here, we report on the use of *E. coli* OverExpress C43(DE3) strain for the expression and production
79 of the *Dd*DyP peroxidase without heme supplement nor its precursor δ-ALA. Furthermore, using X-
80 ray structural analyses, we describe the first crystal structure of native *Dd*DyP peroxidase bound to a
81 natively incorporated heme and demonstrate that the geometry of the heme binding pocket resembles

82 in much detail that of a previously reported cyanide native *Dd*DyP structure, which was prepared
83 following *in vitro* heme reconstitution approach.

2   Methods

2.1   Overexpression and purification of *Dd*DyP with natively incorporated heme

86 The gene sequence encoding dye-decolorizing peroxidase from the slime mold *D. discoideum* AX4
87 (GenBank: EAL70759.1) was codon optimized for *Escherichia coli*, synthesized and subcloned into
88 the *Bam*HI/*Xho*I cloning site in a pGEX-6P1 vector harboring a Human Rhinovirus 3C excision site
89 and a glutathione transferase (GST) tag at the N-terminal region (BioCat GmbH, Germany). The
90 pGEX-6P1-*Dd*DyP construct was transformed into an OverExpress *E. coli* C43(DE3), a chemically
91 competent strain (Sigma-Aldrich, Germany). For purification, a single colony from a freshly streaked
92 plate was inoculated into a Luria-Bertani (LB) or 2×YT enriched media containing 100 µg ml^{-1} final
93 concentration of ampicillin and incubated at 37 °C for overnight (15-16 h). A starter culture was used
94 to inoculate 6× 1L of LB or 2×YT containing 100 µg ml^{-1} ampicillin and incubated at 37 °C or lower
95 temperatures until the optical density (OD$_{600}$) reaches 1.0-1.25 before inducing the expression of
96 *Dd*DyP with 1.0 mM of isopropyl β-D-1-thiogalactopyranoside. The culture was then incubated for
97 additional 7 h or 20 h for expression at 37 and 20 °C, respectively. The cells were harvested by
98 centrifugation with 13,881 ×g for 30 min on an F9-6-×1000 LEX rotor (Thermo Fischer Scientific,
99 Germany) at 4 °C and pellets were stored at -80 °C until used.

100 For protein purification, frozen cells were thawed using warm tab water (~40 °C) and diluted with 3-
101 5× of lysis buffer containing 0.05 M Tris-HCl, pH 8.0 and 0.15 M NaCl supplemented with 1.0 mM
102 final concentration of phenylmethylsulfonyl fluoride protease inhibitor or a tablet of EDTA-free
103 protease inhibitor cocktail (Sigma-Aldrich, Germany). The cells were lysed with 35 cycles of
104 sonication at 4 °C on ice using 50% amplitude and 25 sec sonication pulse with 1.5 min interval. Lysate
105 was clarified with centrifugation at 52,400 ×g at 4 °C for 45 min and the supernatant was filtered with
106 a 0.45 µm syringe filter and mixed with 5-10 ml glutathione sepharose high-performance resin pre-
107 equilibrated with lysis buffer, followed by incubation at 4 °C with gentle rotation for 3 h. The mixture
108 was loaded into an empty gravity column and the GST-tagged *Dd*DyP was eluted with 5× column
109 volume of an elution buffer containing 0.05 M Tris-HCl, pH 8.0, 0.15 M NaCl and 15-20 mM L-
110 Glutathione (reduced form). The GST tag was then removed using HRV 3C protease with 1:20 enzyme
111 to protein ratio at 4 °C for overnight followed by passing the mixture through a glutathione sepharose
112 column pre-equilibrated with lysis buffer. Purified protein was characterized with SDS-PAGE and UV-
113 visible spectrophotometry. For crystallization the protein was further purified with gel-filtration using
114 Superdex 75 10/300 Increase column (Cytiva, Sweden). Purified protein was concentrated to 20-30 mg
115 ml^{-1} in lysis buffer and stored at -80 °C until further use. Overexpression, purification and
116 crystallization were carried out the XBI BioLab of the European XFEL facility (Han et al., 2021).

2.2   Heme reconstitution

118 A control heme reconstitution experiment was conducted as described previously (Chen et al., 2015).
119 In brief, purified apo-*Dd*DyP from LB expression was mixed with hemin chloride with ~1:2 molar
120 ratio in a buffer containing 50 mM Tris-HCl, pH 7.0 and 150 mM NaCl, followed by incubation on ice
121 for 30 min. The heme reconstituted holo-*Dd*DyP protein was then passed through a PD-10 desalting
122 column (Cytiva, Sweden) to remove the excess hemin chloride.

2.3   UV-visible spectrophotometry

3

124 All spectra were recorded on a Shimadzu UV-2700 PC spectrophotometer (Shimadzu Co., Japan) using
125 a cuvette with 1.0 cm pathlength in a range of 200-700 nm at room temperature (20 ± 2.0 °C). For
126 measurements purified DdDyP was diluted with lysis buffer to a concentration of 0.4 mg ml^{-1} and the
127 lysis buffer was used as a reference. All spectra were processed using the Origin software 2022b
128 (OriginLab Co., USA).

129 **2.4 Crystallization screening and crystal optimization of DdDyP**

130 Crystallization screening was performed using a NT8 Formulatrix robot (Formulatrix, USA). Hit was
131 obtained from the C12 condition (20% PEG 6000, 0.1 M HEPES, pH 7.0, 0.01 M ZnCl$_2$) of the
132 PACT++ crystallization screen (Jena Bioscience, Germany) with 10 mg ml^{-1} of purified DdDyP. This
133 condition was further optimized to 15% PEG 6000, 0.1 M HEPES-NaOH, pH 7.0, and 0.01 M ZnCl$_2$
134 crystallized with 15 mg ml^{-1} final concentration of purified DdDyP, which gave rise to a maximum
135 crystal size of 200 × 100 × 25 µm at 20 °C in 4-6 weeks. Crystals were harvested directly from the
136 drops using nylon loops and flash-cooled in liquid nitrogen.

137 **2.5 X-ray diffraction data collection, processing and structure determination**

138 X-ray diffraction datasets were collected at the P11/PETRA III beamline at DESY (Hamburg,
139 Germany) using a flat focus with 20 × 20 µm^2 (v × h) beam area at the sample position, 12.0 keV
140 photon energy, and a photon flux of ~2 × 10^{10} photon sec^{-1} and an exposure time of 100 ms for a total
141 wedge of 360° with 0.1° oscillation recording step on EIGER 16M detector (Burkhardt et al., 2016).
142 Data collection were performed at cryogenic temperature, 100K. Diffraction datasets were processed
143 using the program XDS, and scaled with XSCALE in the XDS graphic user interface (Kabsch, 2010).
144 The initial phase was obtained by molecular replacement using the DdDyP peroxidase active structure
145 (PDB ID: 7ODZ) as a reference model with the program Phaser in the phenix software (Afonine et al.,
146 2012). The model was then refined in phenix and manually corrected in coot (Emsley et al., 2010).
147 Radiation dose was estimated using the program RADDOSE-3D as described previously using the
148 abovementioned parameters (Bury et al., 2018). For channel and cavity calculations *MOLEontile* tool
149 (https://mole.uplo.cz/method) was used with the coordinate obtained from the final cycle of refinement
150 (PDB ID: 8OHY) as a template (Sehnal et al., 2013). The interfaces of heme b and the oligomeric states
151 analysis of DdDyP were calculated using PISA (Protein Interfaces, Surfaces, and Assemblies) server
152 (Krissinel and Henrick, 2007).
153

154 **3 Results and Discussion**

155 **3.1 Heterologous expression and characterization of DdDyP**

156 For heterologous expression, a DdDyP peroxidase gene was cloned into a pGEX-6P1 vector which has
157 an HVR 3C excision site and a GST-tag in its N-terminal region as reported previously (Rai et al.,
158 2021), however, we used the OverExpress *E. coli* C43(DE3) strain instead of BL21 for overexpression
159 screening. DdDyP was expressed at high and low temperature (37 and 20 °C) in LB and 2×YT medium
160 with different yields, ranging from 2.1 and 14.1 mg of protein in average per 25 g of cells, respectively
161 (Table 1). Figure 1A-B shows the SDS-PAGE analyses of the typical expression and purification of
162 DdDyP in *E. coli* C43(DE3) strain. Protein expressed at 37 °C, however, has transparent to pale
163 brownish colour, whereas those of 20 °C exhibited a darker brownish colour (Figure 1C). The proteins
164 purified from high and low temperature have a reasonable purity with less aggregation (Figure 1) with
165 ~35 kDa molecular weight as confirmed by the SDS-PAGE analysis. However, no crystallization hit
166 was obtained from these conditions despite several attempts with various crystallization screens (Table

167   1). Since that our purified *Dd*DyP has sufficient purity for crystallization, yet we did not successfully
168   crystallize it, we concluded that the instability of the protein during expression at 37 °C may be the
169   cause for the unsuccessful crystallization. This is likely due to the improper protein folding, and thus
170   lowering the temperature of the expression may be one method for achieving a stable and correctly
171   folded protein (Francis and Page, 2010; Huang et al., 2021). Indeed, when we expressed *Dd*DyP at 20
172   °C, it gave rise to a darker brownish protein (Figure 1C), a typical heme peroxidase colour of a native
173   protein, especially when expressed in a 2×YT enriched medium, which is richer than LB. Intriguingly,
174   *Dd*DyP expressed in 2×YT at lower temperature was the only condition that resulted in a successful
175   crystallization yielding dark brownish crystals, an indication that heme *b* is preserved during
176   purification and crystallization (Figure 1D). The 2×YT also yielded several times higher amount of
177   protein than that obtained in LB (Table 1).

178   To examine the quality of the electronic absorbance of purified *Dd*DyP we used a UV-visible
179   spectrophotometer. Figure 2 shows the spectral analysis of the purified *Dd*DyP protein from different
180   conditions. *Dd*DyP shows weak absorbance Soret peak at $\lambda = 405$ nm and a Reinheitszahl (R_Z)
181   (A_{soret}/A_{280}) value of ~0.27 when expressed in LB at 37 °C (Table 1). This R_Z value is about 2 times
182   higher than that obtained previously on peroxidases that were expressed using BL21(DE3) strain (Rai
183   et al., 2014; Fiege et al., 2018). As shown in Table 1, we observed that the R_Z value increases ×2–3
184   times to reach ~0.93 with $\lambda = 402$ nm of the Soret peak when expressing *Dd*DyP in enriched 2×YT
185   medium at low temperature (20 °C). This value is comparable to our heme reconstitution reference
186   (Figure 2E) and significantly higher (about 7 times) than those previously reported, when DyP-
187   peroxidases were expressed without adding heme supplements or δ-ALA during expression (Fiege et
188   al., 2018; Krainer et al., 2015). Heme biosynthesis in *E. coli* relies on an L-glutamate amino acid and
189   other proteinoid biochemicals which are abundant in both tryptone and yeast extracts—major
190   components in 2×YT and LB medium (Layer et al., 2010; Krainer et al., 2015). The 2×YT medium has
191   a double amount of tryptone and yeast extracts comparing to LB medium. This might explain the higher
192   level of heme biosynthesis, and thus the native incorporation into *Dd*DyP, in 2×YT than that in LB
193   (Table 1). The heme reconstituted *Dd*DyP from LB expression shows a Soret absorbance at 406 nm
194   and an electron transfer (ET) (*Q*-band) at $\lambda = 497$ nm plus two additional ET bands at 536 nm and 576
195   nm as well as a charge transfer (CT) component at 636 nm, preserving some bacterial peroxidase
196   features (Chen and Li, 2016). This region differs significantly from that previously reported in *Dd*DyP,
197   which showed an ET and CT band at 506 nm and 636 nm, respectively (Rai et al., 2021). We also
198   observed that the *Dd*DyP with a natively incorporated heme has a broad ET peak with $\lambda_{max} = 508$ nm,
199   which is slightly red shifted with $\Delta\lambda = 9$ nm and $\Delta\lambda = 2$ nm, comparing to that of the ET bands of the
200   reference (Figure 2) and a previous work, respectively (Rai et al., 2021). The *Q*-band region also
201   reveals a unique shoulder at the ET band with 567 nm absorbance (Figure 2C).

202   **3.2   Native structure of *Dd*DyP peroxidase bound to a water molecule**

203   Several *Dd*DyP structures have been resolved so far including a single native structure that is in
204   complex with cyanide (PDB: 7O9L) (Rai et al., 2021), however there is no available structure that
205   describes the native resting state. Here, to get insight into the heme binding pocket in its native form,
206   we crystallized the native *Dd*DyP peroxidase bound to a natively incorporated heme and compared it
207   with that resolved in complex with cyanide. A single crystal with a size of $150 \times 80 \times 30$ µm^3 size was
208   used for diffraction data collection (Figure 1D). The crystal data collection and refinement statistics
209   are shown in Table 2. Native *Dd*DyP peroxidase is crystallized in a tetragonal space group $P4_1 2_1 2$
210   similar to previously reported structures (Rai et al., 2014), with exception that the X-ray data of the
211   current crystal condition can be equally processed and resolved in an additional space group ($P4_3 2_1 2$)
212   (Table 2). Moreover, the crystal unit cell exhibited significantly shorter axes, giving rise to about 35%

213   smaller cell volume with 52.8% solvent content and 2.62 Å^3 Da^{-1} of Matthew's coefficient (V_m),
214   indicating the presence of a single molecule per asymmetric unit. The solvent content is decreased by
215   ~20% comparing to that previously reported (Rai et al., 2021). This is more likely due to a relatively
216   high concentration of the dehydrating precipitant (~30% PEG 6000), as previously reported in other
217   systems (Umena et al., 2011; Koua et al., 2013). Such high PEG concentration causes a shrinking in
218   the protein crystals by dehydration which shorten the axes and leads to a tightly packed unit cell (Figure
219   S1). The average radiation dose on a single crystal was estimated with RADDOSE-3D (Bury et al.,
220   2018) to be ~0.58 MGy (Gy = J•kg^{-1}) (Table 2) which lies well below the 20 MGy dose limit suggested
221   by Henderson (Henderson, 1990) or the 30 MGy suggested by Owen and Garman (Owen et al., 2006),
222   indicating that the structure is less affected by radiation damage.

223   The overall architecture of *Dd*DyP is similar to that of the typical DyP-type peroxidase superfamily
224   (Sugano et al., 2007; Chen et al., 2015; Rai et al., 2021). *Dd*DyP contains a duplicated ferredoxin-like
225   fold domain arranged as a *β*-barrel at the N- and C-terminals of the protein (Figure 3). It contains 12
226   *β*-sheets and 13 *α*-helices formed by 185 residues of the full chain (306 residues), and the remaining
227   121 residues involved in the formation of loop structures that link these secondary structures. Similar
228   to all other known DyP-type peroxidases, *Dd*DyP contains *α*-helices with a unique β-sheet structure at
229   the distal region of the heme plane (Strittmatter et al., 2013; Sugano et al., 2007). We determined the
230   root mean square deviation (r.m.s.d.) between the C*α* (1-306 residues) of the present structure with that
231   resolved in complex with cyanide (PDB ID: 7O9L) to be 0.18 Å, indicating the striking similarity
232   between the two structures. Our PISA analysis predicted a stable dimer of *Dd*DyP in solution with 33
233   residues contributing to the dimer interface, similar to previously reported *Dd*DyP structures (Rai et
234   al., 2021). These interfacial residues are distributed along the dimer interface from the N- to the C-
235   terminal region. The dimeric structure reveals a solvent accessible area of 24290 Å^2 and buried surface
236   areas (BSA) of 5330 Å^2, corresponding to about 22% of the total surface area of the protein. On the
237   other hand, the BSA of the monomeric structure is 1341 Å^2, corresponding to 9.6% of the total surface
238   area of monomeric *Dd*DyP. It should be noted that our PISA analysis favoured a tetramer oligomeric
239   state for the native cyanide *Dd*DyP (PDB ID: 7O9L) structure, displaying higher binding energy than
240   that of the dimeric state. This indicates that *Dd*DyP protein may exist physiologically in various
241   oligomeric states. Indeed, several DyP-type peroxidases have been reported to exist in different
242   functional oligomeric states ranging from monomeric to tetrameric state (Yoshida et al., 2016; Pfanzagl
243   et al., 2020; Liu et al., 2011; Zubieta et al., 2007). The catalytic arginine residue, Arg239 in *Dd*DyP,
244   has been suggested to play a role in the protein oligomerization owing to its location and hydrogen
245   bonding network with surface residues (Chen et al., 2015; Singh et al., 2012). In *Dd*DyP, Arg239 is
246   buried in the hydrophobic cavity of the heme binding pocket, excluding its contribution in *Dd*DyP
247   oligomerization. Moreover, our molecular replacement attempts aiming for a dimeric solution was not
248   successful, thus we can reasonably conclude that our purified *Dd*DyP favours a monomeric state in
249   crystal. It has been previously reported, based on sedimentation velocity analysis with analytical
250   ultracentrifugation, that dimeric *Dd*DyP predominates in solution, which yielded a dimeric crystal
251   structure (Rai et al., 2021). The crystal packing behaviour of the present structure (PDB ID: 8OHY) is
252   significantly different from that described previously (Rai et al., 2021), likely due to a significantly
253   low unit cell volume which exerts tight interactions between molecules in the unit cell (Figure S1).
254   Note that the molecular contact within the unit cell is contributed by similar regions in both forms that
255   is primarily the loop and β-sheet of the ferredoxin-fold like domain II at the C-terminal region (Figure
256   S1).

257   **3.3**   **Geometry of a natively incorporated heme, its binding pocket and the implication in**
258   **catalysis**

259   Heme *b* in the DyP-type peroxidases, a protoheme IX, is either penta- or hexacoordinated (Sugano et
260   al., 2007; Singh et al., 2012; Rodrigues et al., 2021). The native structure of *Dd*DyP accommodates
261   heme *b* in a hydrophobic binding pocket flanked by the unique β-sheet at the distal side of the heme
262   plane, the α-helices of the ferredoxin-like fold domain II (Figure 4) and a distinct long loop at its
263   proximal side similar to previously reported DyP structures (Sugano et al., 2007; Zubieta et al., 2007;
264   Liu et al., 2011). Our structural analysis shows that the native *Dd*DyP heme is hexacoordinated, of
265   which the pyrrole rings of porphyrin contributed to tetradentate chelation via their nitrogen atoms and
266   via the conserved His222 at the proximal side with a distance of 2.11 Å. The sixth coordination is
267   provided by a water molecule (wat-184) with a distance of 2.79 Å, which is shorter by ~0.1 Å than that
268   of the reported $Fe(III)\!-\!CN$ distance, indicating a stronger coordination to Fe(III) (Figure 4C). This
269   distance is typical for Fe(III) of the resting state, implying that the model is unaltered by radiation
270   damage (Chen et al., 2015). Wat-184 forms a strong hydrogen bond (~2.2 Å) with wat-182 and a
271   weaker hydrogen bond with the catalytic residue Arg239 (Figure 4C). Intriguingly, the environment of
272   our native *Dd*DyP heme binding pocket is similar to that of the bacterial DypB peroxidases (Chen et
273   al., 2015). We observed that wat-184 has slightly higher *B*-factor than wat-182, which may indicate its
274   mobility and higher reactivity. On the other hand, wat-182 interacts via strong hydrogen bonds with
275   the second catalytic residue Asp149 as well as Ser241, suggesting that these residues may act as proton
276   acceptors to the H_2O_2 during the formation of compound I oxyferryl thereby contributing to its
277   stabilization along with Arg239 (Figure 4C) as revealed in other A-type DyP peroxidases (Pfanzagl et
278   al., 2018). The binding pocket is extensively occupied with water molecules which are in hydrogen
279   bonding interaction with nearby residues that contribute to the heme stability (Figure 4C). Note that
280   two of the water molecules (wat-150 and wat-159), which are in hydrogen bonding interaction with
281   Asp149 and Arg137 near the heme access channel, substituted the 1,2-ethanediol molecule in the
282   cyanide native structure that shifted Asp149 carboxylate group towards the cyanide, giving rise to
283   increased *B*-factor of Asp149 comparing to its surrounding (Rai et al., 2021). This may indicate that
284   this position is natively occupied by water molecules as demonstrated by our native structure.

285   The heme ligand is well resolved at a resolution of 1.95 Å as revealed by its 2*m*Fo–*D*Fc electron density
286   map (Figure 4B and Figure S2), indicating unambiguous incorporation and binding of the heme in the
287   apo-protein. This is an important finding as crystallization with purified *Dd*DyP proteins that have
288   lower R_Z values were not successful, which might indicate that the heme on these purified proteins is
289   not well accommodated in the binding pocket, affecting possibly their stability and hence the
290   crystallization (see Table 1). The heme occupies 798 Å^2 surface area, corresponding to 5.7% of the
291   total surface area of the native structure, similar to that of the cyanide native *Dd*DyP structure and other
292   *Dd*DyP structures (Rai et al., 2021). PISA analysis indicates that the solvation free energy gain ($\Delta^i G$)
293   of the natively incorporated heme is -22.5 kcal mol^{-1} with 622 Å^2 interface area comparing to an
294   average of -22.3 kcal mol^{-1} for the reconstituted heme of the cyanide native structure (PDB ID: 7O9L)
295   which has an interface interaction area of 609 Å^2, indicating similar heme binding affinity with slightly
296   better properties for the natively incorporated heme. The van der Waals interactions as well as the
297   hydrogen-bonding network provided by nearby residues and water molecules may contribute to the
298   stabilization of heme binding (Figure 4C) (Sacquin-Mora and Lavery, 2006; Mogharrab et al., 2007).
299   Our analysis shows that the heme is stabilized, along the plane, via its carboxylate oxygens by hydrogen
300   bonding with several water molecules (wat-120, wat-79 and wat-166), and three residues, Glu152,
301   Arg204 and Arg239. The Arg239 interacts weakly with the heme carboxylates oxygens via two
302   hydrogen bonds with a distance of 3.3-3.4 Å, whereas Glu152 and Arg204 form hydrogen bonding
303   with the heme via wat-120 and wat-166, respectively. These interactions indicate that the heme is well
304   stabilized in our model (PDB ID: 8OHY), which confirms the correct geometry of its native
305   incorporation, yielding comparable binding pocket geometry to that prepared with *in vitro*
306   reconstitution (Chen et al., 2015; Rai et al., 2021). Furthermore, the superimposition with the cyanide

307   native structure indicates a striking similarity (r.m.s.d. = ~0.18 Å) around the heme binding pocket
308   including the flanking loop at the proximal side of the heme plane (residues 204-220) (Figure 3D). A
309   side-specific mutagenesis study in DypB found that this proximal loop may have significant role in the
310   heme stability (Rodrigues et al., 2021). This loop has also been implicated in the stabilization of the
311   substrate owing to its flexibility thereby facilitating the substrate/product turnover by flipping in and
312   out around the heme binding site (Liu et al., 2011).

313   Further, we used the *MOLEonline* tool (https://mole.upol.cz/online) to analyse the cavities and tunnels
314   nearby the heme binding pocket and those in long-range distances (Figure S3) (Sehnal et al., 2013).
315   Overall, 14 tunnels were identified, three of which are located next to the heme and perpendicular to
316   each other with characteristics that might have an implication in the *Dd*DyP catalysis—entry of
317   substrates and exit of reaction products. These channels may serve as entry gates for H_2O_2 thereby
318   facilitating the enzyme activation required for the oxidative catalysis (Chen et al., 2015; Yoshida et al.,
319   2016; Habib et al., 2019). Two tunnels have average diameter of ~3.0 Å, which is sufficient to facilitate
320   the entry of H_2O_2 and perhaps the exit of reaction products of similar size. All tunnels are lined with
321   hydrophobic residues in the middle of the channel as well as several key catalytic residues in the distal
322   and proximal sides of the heme plane. In particular, Arg239, Asp149, His222, Glu152, Ser241, and
323   Thr226 in addition to several hydrophobic residues where found in two proximal channels (Figure S3),
324   which are converged to form a main channel with a length of 42 Å and a diameter of ~4.0 Å. The
325   access gate of this channel is lined with charged residues as shown in Figure 4D and E, indicating its
326   implication in the substrate entry. Further, we identified two major cavities at the distal side of the
327   heme plane, of which one cavity (cavity 1) has a volume of 2881 $Å^3$, corresponding to 9.6% of the total
328   surface of *Dd*DyP and 4.7 times of the heme molecule. It is located at the heme binding pocket,
329   accommodating the main channel at the binding pocket and extends to the proximal side of the heme
330   plane, indicating a role for this cavity in the catalysis of DyP-peroxidases (Rai et al., 2021; Habib et
331   al., 2019; Yoshida et al., 2011). The second cavity (cavity 2) with approximately half a volume of that
332   of cavity 1 (1411 $Å^3$) is located at the N-terminal region distant from the heme binding pocket and in
333   contact with cavity 2 near the molecular centre of *Dd*DyP (Figure S3). The presence of such cavities
334   is important for accommodating wide-range of substrates thereby fulfilling the substrate broad
335   specificity of DyP-type peroxidases (Pfanzagl et al., 2019; Silva et al., 2023).

4   Conclusions

337   In conclusion, we demonstrated the use of *E. coli* C43(DE3) strain for heterologous expression of
338   *Dd*DyP peroxidase, without the use of a heme precursor *δ*-ALA, hemin chloride or iron supplement,
339   to produce to produce *Dd*DyP holoprotein with a natively incorporated heme, relying primarily on the
340   *E. coli* heme biosynthesis by benefiting from the use of enriched medium and low temperature during
341   expression, which yielded an R_Z value of ~1.0 and a holoprotein with sufficient stability. We further
342   showed, by mean of X-ray crystallography, that the native *Dd*DyP expressed in this condition has
343   comparable heme geometry and binding properties. Our study also demonstrates that the natively
344   incorporated heme is well stabilized via hydrogen bonds provided by nearby Arg239, Glu152 and water
345   molecules in addition to van der Waals interactions between the porphyrin rings and surrounding
346   residues within van der Waals distances. Two cavities occupying a total volume of 4292 $Å^3$,
347   corresponding to 14.3% of the total monomeric volume (29951 $Å^3$), were identified. Of which the main
348   cavity (cavity 1) around the heme binding pocket was found to accommodate a large access channel
349   that spans the heme binding pocket. Moreover, the high-quality crystals optimized in this work would
350   be suitable for use as a model for metalloenzymes to study the dynamics and substrate binding kinetics
351   during catalysis. This can be achieved by, for instance, the mixing-and-inject time-resolved serial
352   femtosecond crystallography approach (Pandey et al., 2021), which enables tracking the formation of

353 the reaction intermediates as well as the mechanism of substrate breakdown into products as
354 demonstrated in other metalloproteins (Worrall and Hough, 2022; Malla and Schmidt, 2022).

355 **5   Conflict of Interest**

356 *The authors declare that the research was conducted in the absence of any commercial or financial*
357 *relationships that could be construed as a potential conflict of interest.*

358 **6   Author Contributions**

359 Conceptualization, F.H.M.K.; experiments, Ö.K., L.B., H.H., and F.H.M.K.; formal analysis and data
360 interpretation, F.H.M.K.; Diffraction data collection and processing, F.H.M.K., Ö.K., and S.K.; X-ray
361 structural analysis, F.H.M.K.; contributed funding/reagents/analytic tools, A.P.M; supervised the
362 project, F.H.M.K.; manuscript writing, F.H.M.K. with input from all authors. All authors read and
363 approved the final version of this manuscript.

364 **7   Funding**

368 **8   Data availability statement**

369 The crystal structure data was deposited to the protein data bank under the PDB ID: 8OHY and is
370 publicly available under the PDB digital object identifier http://doi.org/10.2210/pdb8OHY/pdb.

371 **9   Acknowledgments**

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381 **10   References**

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547 **TABLES**

548 **Table 1** Characterization of the native heme incorporation into *Dd*DyP expressed in *E. coli* C43(DE3)
549 at different conditions, the protein yield and crystallization trials

Overexpression	Soret peak (nm)	R_Z value (A_{soret}/A_{280})	Heme content (%)[†]	Protein yield (mg)[‡]	Crystallization hits
LB at 37 °C	406	0.27	26.5	1.2	No
LB at 20 °C	405	0.502	49.3	2.96	No
2×YT at 20 °C	402	0.93	90	14.1	Yes
Reference*	406	1.02	100	1.2	Not tested

550 *The reference is the *Dd*DyP with a reconstituted heme; from LB expression. [†]Heme content relevant
551 to the reference value in this study, which was set to 100%. [‡]The yield is normalized to 25.0 g of wet
552 weight of overexpressed cells used for purification.

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568 **Table 2** X-ray diffraction data collection and crystallography refinement statistics

	Native *Dd*DyP structure
PDB ID	8OHY
Data Collection	
Beamline	P11/PETRA III at DESY
Photon Energy (keV)	12.0
Photon flux (ph . s^{-1})	~2 × 10^{10}
Radiation dose (MGy)	~0.58
Space group	$P4_1 2_1 2$
Cell dimensions	
a, b, c (Å)	99.88 99.88 73.12
α, β, γ (°)	90.00 90.00 90.00
Resolution (Å)	44.67–1.95 (2.02–1.95)*
R_{merge}†	0.2715 (4.643)*
$I/\sigma I$	12.02 (0.6)*
Completeness (%)	98.54 (91.55)*
Multiplicity	26.7 (26.0)*
$CC_{1/2}$	0.999 (0.411)*
CC^*	1.0 (0.764)*
Wilson B-factor (Å^2)	36.8
Refinement	
Resolution range (Å)	44.67–1.95
No. of reflections (unique)	27129 (2471)*
R_{work} / R_{free}‡	0.206 (0.400)* / 0.247 (0.445)*
No. of atoms	2662
Protein	2459
Ligands	74
Solvent	163
No. of residues	306
Average B-factor (Å^2)	41.29
Protein	41.23
Ligands	39.63
Solvents	42.76
r.m.s. deviations	
Bonds (Å)	0.005
Angles (°)	0.71
Ramachandran (%)	
Favored	98.03
Allowed	1.97
Outliers	0.00
Rotamer outliers (%)	0.00
Clashscore	1.00

569   *Values in parenthesis are of the highest resolution shell. $^†R_{merge} = \sum_h \sum_i |I_i(h) - \langle I(h) \rangle| /$
570   $\sum_h \sum_i |I_i(h)|$, where $I_i(h)$ is the intensity measurement for a reflection h and $\langle I(h) \rangle$ is the mean
571   intensity for this reflection. $^‡R_{work} = \sum_h \| F_{obs}| - |F_{calc} \| / \sum_h |F_{obs}|$ and R_{free} was calculated
572   using a randomly (5.0%) selected reflections.
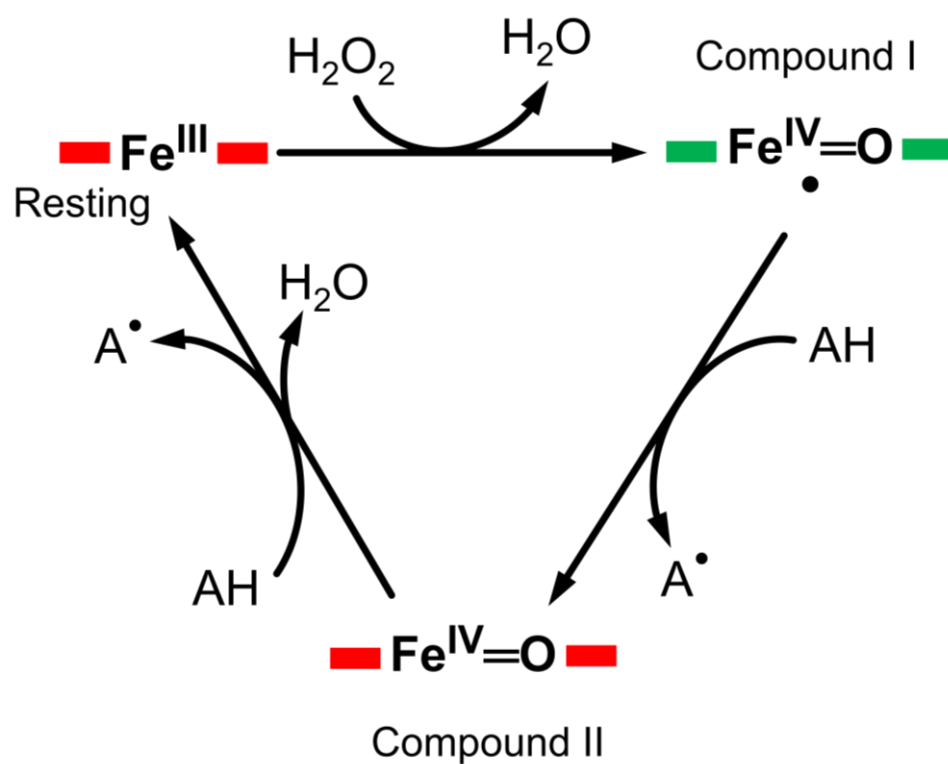
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SCHEME AND FIGURE LEGENDS

Scheme 1 A typical enzymatic cycle of DyP-type peroxidases showing the interconversion between the resting state (ferric porphyrin), compound I heme oxoferryl species (porphyrin cationic radical) $[\text{Fe(IV)}{=}\text{O Por}^\bullet]^+$; and compound II intermediate state, $[\text{Fe(IV)}{=}\text{O}]^+$. The AH is the reducing substrate which is oxidized into an intermediate radical product ($\text{A}^\bullet$) during catalysis.

Figure 1 Purification, characterization and crystallization of *Dd*DyP. (A) An SDS-PAGE showing the glutathione sepharose affinity purification. Lane M, protein marker; FT lane, flow-through of the unbound proteins from the lysate; W-1 lane, step 1 washing of the Glutathione Sepharose column; and W-2, the second washing step of the column after elution (B) the HRV 3C digestion of GST-*Dd*DyP expressed protein complex and GST-free *Dd*DyP protein. (C) The Gel-filtration profile of purified *Dd*DyPs and representative samples from 2×YT expressed at 37 and 20 °C, and LB at 37 °C, and (D) brownish crystals from two different purification batches of independently overexpressed *Dd*DyP protein. The scale bar in "D" represents 100 µm.

Figure 2 UV-visible electronic absorbance of *Dd*DyP of different overexpression conditions. (A) a typical spectrum of a purified *Dd*DyP when expressed in LB medium at 37 °C, (B) and (C) *Dy*DyP electronic spectra when expressed at 20 °C in LB and 2×YT medium, respectively. The ET *Q*-band and charge transfer band in the region from 500–700 nm of the 2×YT condition are described in (C). (D) The reference UV-visible spectra of LB *Dd*DyP after reconstitution with hemin chloride, and the small box in (D) represents a magnified view of the *Q*-band and charge transfer regions from 500–685 nm.

Figure 3 Overall structure of native *Dd*DyP and its superposition with native cyanide *Dd*DyP structure (PDB code: 7O9L). (A) Topology representation of the *Dd*DyP-type peroxidase. (B) Overall structure of monomeric *Dd*DyP showing the ferredoxin-like folds at the N- and C-terminals colored according to (A). (C) Superimposition of native *Dd*DyP structure (green) into a cyanide native structure (gray). (D) Electrostatic potentials surface of native *Dd*DyP colored from -5.7 kT (red) to +5.7 kT (blue) calculated using the program PyMOL (http://www.pymol.org/pymol). The yellow dashed circle highlights the heme binding pocket and possible pathway for H_2O_2 and/or substrate entry.

Figure 4 Monomeric structure of the native *Dd*DyP highlighting the heme binding pocket at a resolution of 1.95 Å (PDB code: 8OHY) and the channel characteristics of the system. (A) The overall monomeric structure of *Dd*DyP highlighting its heme binding pocket (black box). (B) The 2*m*Fo–*D*Fc electron density map (gray mesh) contoured at 1.0 sigma level displaying a magnified view of the heme binding pocket (active site). (C) The hydrogen bonding network (black dots) of the heme binding pocket at the distal and proximal sites of the heme plane. The two water molecules (wat-182 and wat-184) liganded to the heme-Fe(III) via hydrogen bond are highlighted in black, while the conserved key residues are displayed as yellow stick. All figures were generated using PyMOL software (http://www.pymol.org/pymol). (D) Two-dimensional representation of a large pore spanning the C-terminal region across the distal site of the heme binding pocket as a result of two channel convergence colored according to the hydropathy scores of the lining residues. (E) A 3D representation of the channel shown on (D).
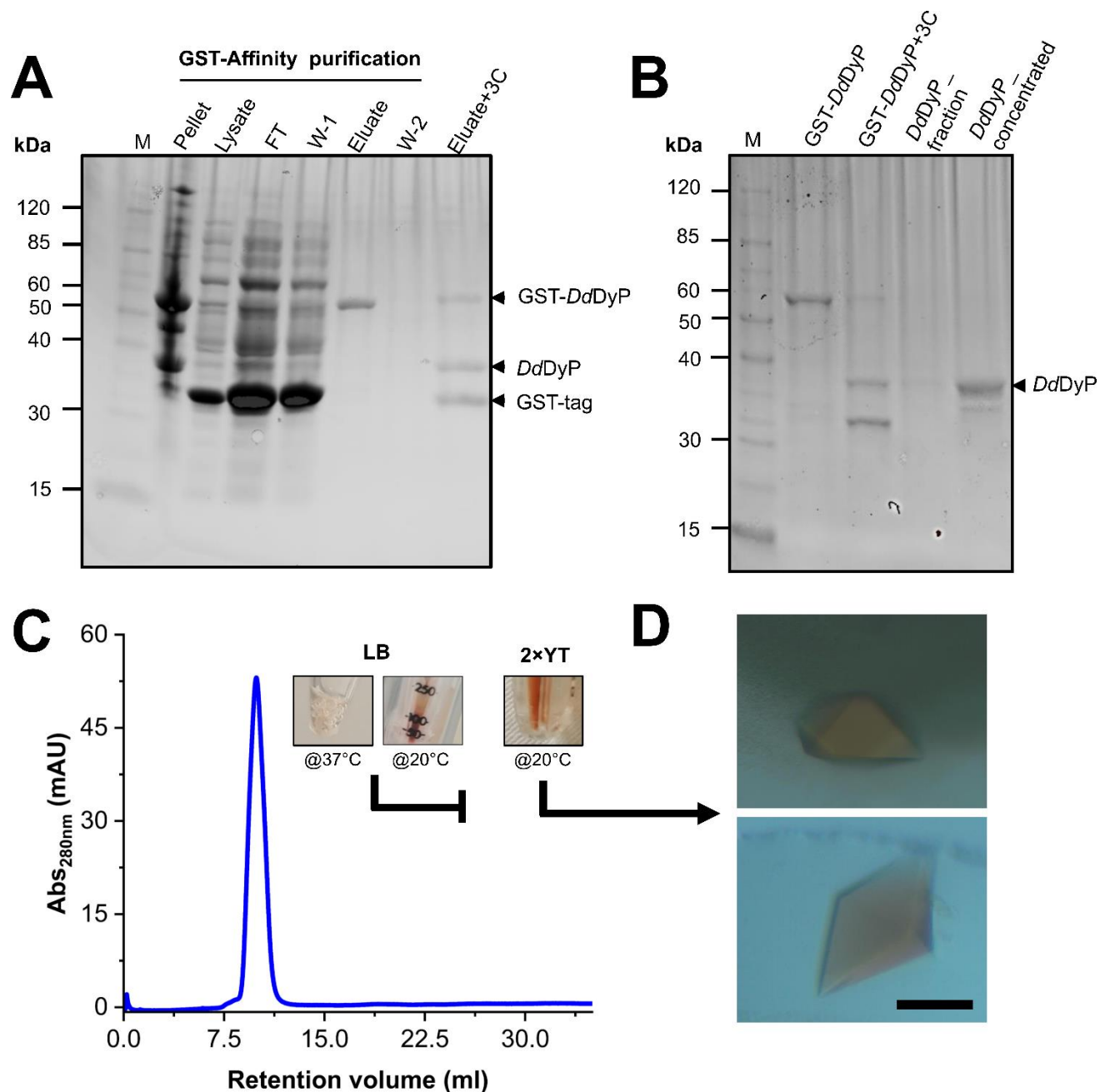
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Figure 1 Kalkan et al

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Figure 2 Kalkan et al

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A

ferredoxin-like fold domain II

ferredoxin-like fold domain I

B

C

D

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649 **Figure 3** Kalkan et al

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Figure 4 Kalkan et al

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